

# Phase Transition in TlS, TlSe and TlInS<sub>2</sub> Crystals Caused by Nanoscale Defects

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**Abstract** - In this paper, we report the results of our studies on temperature and electric field dependences of conductivity in TlS, TlSe and TlInS<sub>2</sub> crystals. At a certain temperature (typical for all three crystals) a switching effect has been observed for the conductivities of all crystals. The switching effect is explained by the phase transition to superionic conductivity. It is suggested that in TlS and TlSe crystals the ion conductivity is caused by the diffusion of Tl<sup>+</sup> ions over vacancies in the thallium sub-lattice between (Tl<sup>3+</sup>S<sup>2-</sup>) and (Tl<sup>3+</sup>Se<sup>2-</sup>) chains. In TlInS<sub>2</sub> crystals, this effect is due the diffusion of Tl<sup>+</sup> ions towards vacancies in the Tl sub-lattice. In all crystals the S-type switching effect is revealed. It is suggested that the switching effect is related to the transition of crystals to the superionic state, which is accompanied by diffusion of Tl<sup>+</sup> ions.

**Keywords:** Superionic conductivity, Crystal, Phase transition, Nanoscale, Low-dimensional, Switching effect.

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## 1. Introduction

Materials with quasi low-dimensional crystalline structure, have been objects of intensive study due their promising physical properties caused by nanoscale effects. These materials have been largely used in the fabrication of miniature accumulators, ionistors, gas sensors, solid-state fuel elements, and other devices.

Superionic conductivity is one of the unique phenomenon observed in quasi low-dimensional crystals. Superionic conductivity exists under the following conditions: the network of channels in the crystal's structure must be strong enough to withstand moving ions; the energy of ion disordering over lattice positions and the energy lost during motion should be small; the number of potentially mobile ions per unit cell should exceed the number of immobile ions [1]. These conditions are satisfied for only a few crystals, the structure of which excludes long-range ordering in the spatial arrangement of one or several types of atoms but retains long-range order for other particles. These compounds are considered crystals with intrinsic structural disordering [2]-[5]. Crystals with structural disordering, which have mainly ion conductivity, can be in two qualitatively different (dielectric and superionic) states. At temperatures below critical, they behave like conventional ionic crystals (dielectric phase), while at temperatures above critical they pass into a peculiar superionic state (electrolytic phase). This effect was studied in TlInSe<sub>2</sub> and TlInTe<sub>2</sub> crystals [6]. It was observed a jump in the dependences  $\sigma(T)$  for both crystals at some critical temperatures (in the range of 333 K – 391 K, varying with respect to the crystals and crystal axis), where  $\sigma$  is conductivity, and  $T$  is the absolute temperature. For both crystals, in the range of the sharp jump the dependence  $\ln(\sigma T)$  vs.  $10^3/T$  was linear. The observed jumps in the conductivity was explained by the sharp change in the number of ions in

the states characterized by high ion mobility, i.e., by the phase transition to the superionic state [3], [6], [7].

TlS, TlSe and TlInS<sub>2</sub> belong to this class of materials, in which the features of low dimensional systems manifest themselves under certain conditions [1]. These crystals are attracting attention because of the features of their crystal structure, more specifically, the pronounced chain (TlS and TlSe) and layered (TlInS<sub>2</sub>) structures [8]. Weak links between chains (in TlS and TlSe) and layers (in TlInS<sub>2</sub>) result in these structures being inclined to nanoscale defects. For example, even in single crystals of this class of compounds, the density of uncontrolled defects can reach 10<sup>20</sup> cm<sup>-3</sup>. In this case, crystals exhibit hopping conductivity similar to that observed in amorphous or highly disordered crystals, which is well described within the Mott approximation.

TlS and TlSe crystals belong to compounds of the A<sup>3</sup>B<sup>6</sup> group, which crystallize into tetragonal system with the space group D<sup>18</sup><sub>4h</sub>. A characteristic feature of these crystals is that their structure is formed by (Tl<sup>3+</sup>S<sub>2</sub><sup>2-</sup>)<sup>-</sup> and (Tl<sup>3+</sup>Se<sup>2-</sup>)<sup>-</sup> chains, elongated along the tetragonal axis c of the crystal (Fig. 1.2, [1]). The tetragonal axis is an optical axis. Monovalent Tl<sup>+</sup> atoms are in the octahedral environment of S and Se atoms in the TlS and TlSe crystals, respectively. Based on the crystal-chemical considerations, it can be suggested that this crystal structure is most favourable for mobile Tl<sup>+</sup> ions. In this case, the favourable factor is the presence of extensive cavities, which are linked by shared faces (conductivity windows), as well as the fundamental possibility of deficit of monovalent thallium ions, as a result of which the ionic conductivity may significantly rise.

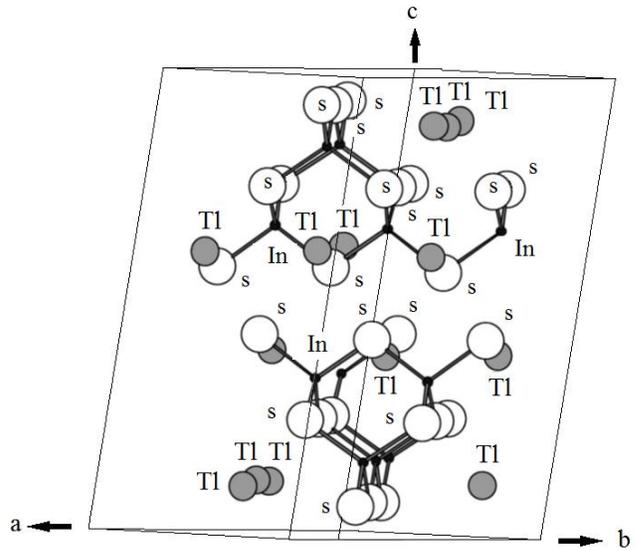


Figure 1. Crystalline structure of TlInS<sub>2</sub>.

TlInS<sub>2</sub> is a *p*-type semiconductor with a layers structure made of In<sub>4</sub>S<sub>10</sub> (which in turn consist of four tetrahedrons InS<sub>4</sub>) groups that form layers extending perpendicular to the c-axis of the material (Fig.1). These negatively charged layers are bonded together by Tl<sup>+</sup> ions. The resulting monoclinic lattice is characterized by group symmetry. Significant attention has been paid to such crystallographic systems, that behave as if they have less than three spatial dimensions [1]-[5], [9], [10]. Such materials are often called quasi-two-dimensional (2D) layers.

In this paper, we report the results of experimental study of conductivity and switching effects in TlS, TlSe and TlInS<sub>2</sub> crystals in a wide temperature range.

## 2. Material and Method

### 2.1. Material

We used Tl and S components for synthesizing TlS; Tl and Se components for synthesizing TlSe; Tl, In and S components for synthesizing TlInS<sub>2</sub> compounds. All components had the purity no less than 99.99%. The corresponding components in the required ratio of masses for each compound were placed in evacuated quartz cells.

### 2.2. Method

The single crystals were grown by the modified Bridgman method [11]. The tetragonal axis c of the freshly cleaved rectangular crystal samples prepared

for study was oriented in the cleavage plane. The conductivity was measured by the four-contact method in two directions: parallel and perpendicular to the tetragonal axis  $c$  of the crystal. The experimental samples were prepared in the form of rectangular plates, 0.4–0.6 mm thick. Contacts with the samples were formed by a silver conducting paste on the plate surface. The permittivity and conductivity were measured by E7-25, digital immittance meters at frequencies of 20–10<sup>6</sup> Hz in a temperature range of 100–450 K. The measuring field amplitude did not exceed 1 V cm<sup>-1</sup>. The temperature dependences of conductivity  $\sigma(T)$ , in TlS, TlSe, and TlInS<sub>2</sub> are shown in Figs. 2, 3 and 4 respectively. The measurements were performed in an electric field oriented perpendicular to the tetragonal crystal axis  $\sigma_{\perp}(T)$  and parallel to it  $\sigma_{\parallel}(T)$ .

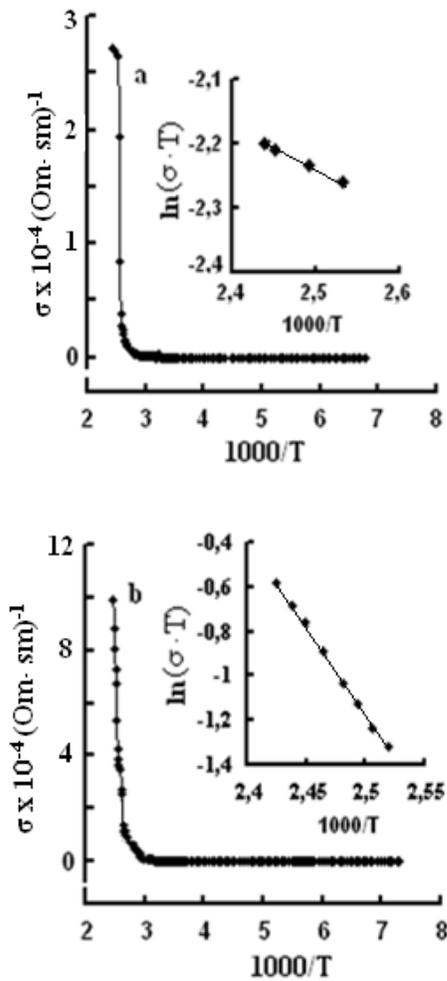


Figure 2. Temperature dependence of conductivity of TlS parallel (on the top) and perpendicular (on the bottom) to axis  $c$ .

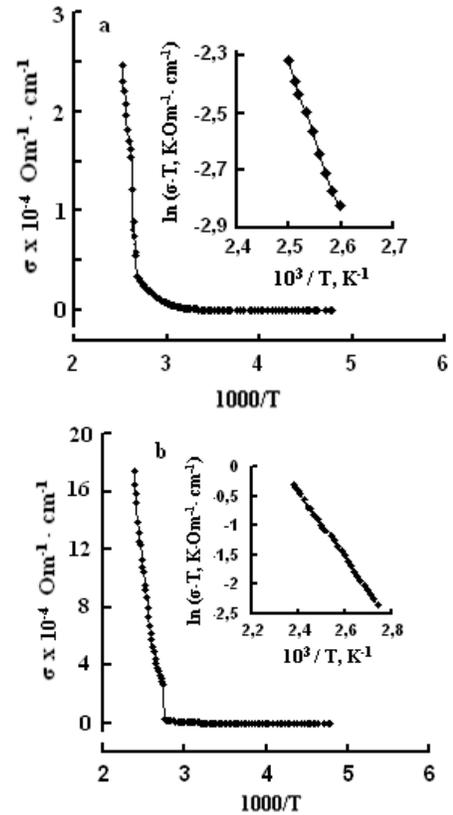


Figure 3. Temperature dependence of conductivity of TlSe parallel (on the top) and perpendicular (on the bottom) to axis  $c$ .

One can see jumps in the dependences  $\sigma(T)$  of all three crystals: for TlS a jump in  $\sigma_{\parallel}(T)$  is observed at a temperature of  $T_{\parallel cr} = 385$  K and a jump on  $\sigma(T)$  at  $T_{\perp cr} = 387$  K; for TlSe the corresponding values are  $T_{\parallel cr} = 367$  K and  $T_{\perp cr} = 380$  K; for TlInS<sub>2</sub> the corresponding value is  $T_{\perp cr} = 390$  K.

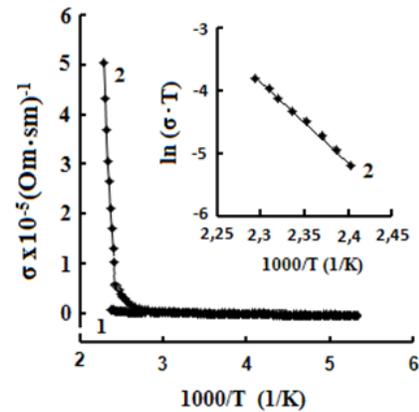


Figure 4. Temperature dependence of conductivity of TlInS<sub>2</sub> parallel (curve 1) and perpendicular (curve 2) to axis  $c$ .

The dependences of conductivities of TlS, TlSe, and TlInS<sub>2</sub> crystals on the electric field strength  $E$  at different temperatures are shown in Fig. 5. The measurements were performed along the tetragonal crystal axis  $c$  and perpendicular to it.

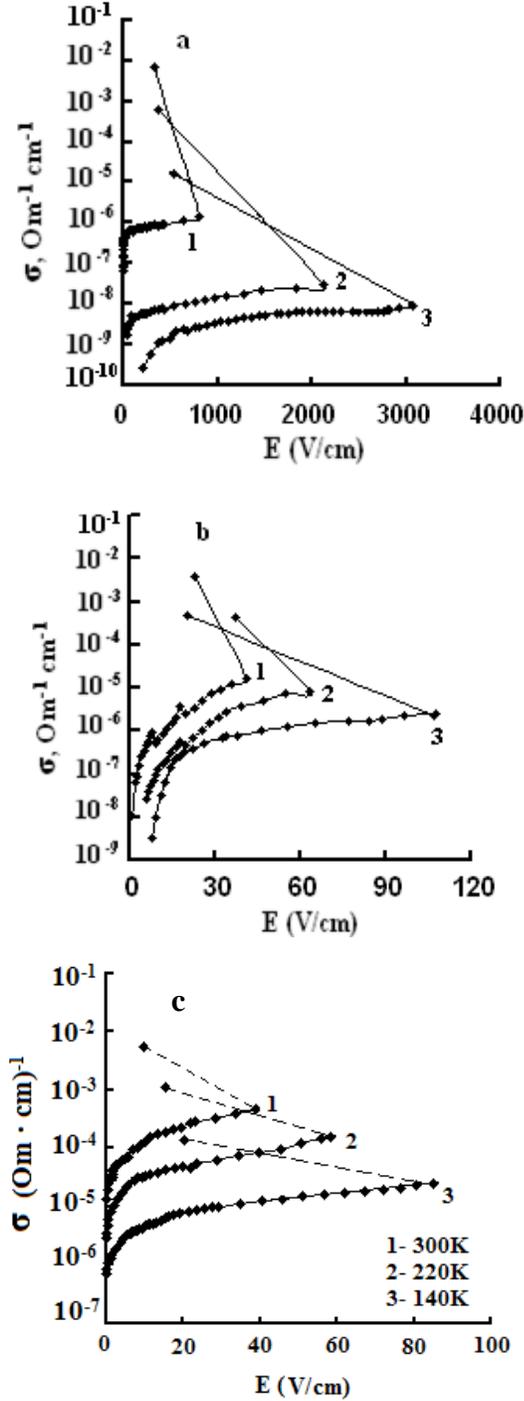


Figure 5. Field dependences of conductivity for a) TlS, b) TlSe and c) TlInS<sub>2</sub>.

## 2.1. Discussions

As follows from the insets of Figs. 2, 3 and 4, for all crystals, the experimental points of the temperature dependence  $\ln(\sigma \cdot T)$  in the range of the sharp jump of conductivity are described well by a straight line, which is given by the following equation:

$$\sigma \cdot T = \sigma_0 \exp\left(-\frac{\Delta E}{kT}\right) \quad (1)$$

where  $\sigma$  is the conductivity of sample,  $\sigma_0$  is the constant within the experimental interval of temperature,  $\Delta E$  is the conductivity activation energy,  $T$  is the temperature in Kelvins and  $k$  is the Boltzmann constant. It is known that this temperature dependence of conductivity indicates the dominant character of ionic conductivity above the critical temperature [2-5]. The values of critical temperatures and activation energies for TlS, TlSe, and TlInS<sub>2</sub> crystals are shown in Table 1.

Table 1. Critical temperatures and activation energies for TlS, TlSe and TlInS<sub>2</sub>.

Crystals	$T_{cr}^{\parallel}$ , K	$T_{cr}^{\perp}$ , K	$\Delta E_{\parallel}$ , eV	$\Delta E_{\perp}$ , eV
TlS	385	388	0.07	0.05
TlSe	367	380	0.1	0.07
TlInS <sub>2</sub>	N/A	390		0.08

The observed jumps in conductivity at critical temperatures can be explained by the sharp change in the number of ions in the states characterized by high ion mobility, i.e., by the phase transition to the superionic state. The linear character of the dependence  $\ln(\sigma \cdot T)$  on  $1/T$  above the critical temperature indicates dominance of ionic conductivity, which was clearly reasoned in [2]-[5]. This reasoning can also be strengthened by considering the structural aspect, i.e. by the existence of channels of conductivity along the chains and layers. It is suggested that in TlS and TlSe crystals the ion conductivity is caused by the diffusion of Tl<sup>+</sup> ions over vacancies in the thallium sub-lattice between (Tl<sup>3+</sup>S<sub>2</sub><sup>2-</sup>) and (Tl<sup>3+</sup>Se<sub>2</sub><sup>2-</sup>) chains. In TlInS<sub>2</sub> crystals, this effect is due the diffusion of the Tl<sup>+</sup> ions towards vacancies in the Tl sub-lattice between In<sub>4</sub>S<sub>10</sub> tetrahedral complexes (fig. 1). Therefore, the observed jump in the conductivity, the linear character of the dependence  $\ln(\sigma \cdot T)$  vs.  $1/T$ , along with the structural factor, allow us to confirm the existence of superionic conductivity in these crystals.

In all crystals, the S-type switching effect is revealed. It is suggested that the switching effect is related to the transition of crystals to the superionic state, which is accompanied by diffusion of Tl<sup>+</sup> ions. This conduction mechanism is typical of superionic conductors [6], [9], [12], [13]. The transition to the high-conductivity state in superionic conductors occurs generally as a result of the first order phase transition and is explained by the stepwise disordering of one of the crystal sub-lattices (generally cationic) with the other sub-lattice either remaining invariable or transforming but retaining the crystal hardness. This effect was found experimentally in  $\alpha$ -AgSbS<sub>2</sub>, where application of an external electric field caused gradual increase in conductivity with a subsequent sharp increase when the field reached the critical value [10].

As one can see from the field dependences of the conductivity (Fig. 5), in relatively weak fields, conductivity  $\sigma$  is almost independent of the applied field  $E$  because of the dominance of the electronic component in  $\sigma$  in this range of field strengths. A further increase in  $E$  led to a linear increase in  $\sigma$ , which is explained by the increase in the ionic component of conductivity as a result of gradual ordering of the cationic Tl sub-lattice in the electric field; in this range, the ionic conductivity begins to dominate over the electronic component and, when reaching the critical temperature, the conductivity sharply rises.

The values of the critical transition field and the jump of conductivity during the phase transition increase with a decrease in temperature (Fig. 5). Voltage oscillations were observed in the range of negative differential resistance in both compounds; these were also revealed in TlInS<sub>2</sub>.

In our opinion, the discovered effect of S type switching in the TlS, TlSe and TlInS<sub>2</sub> crystals, as well as the voltage oscillations in the range of negative differential resistance, are related to the transition of these crystals to the superionic state, which is accompanied by melting of the Tl sub-lattice.

The measured dependences of the TlS and TlSe conductivities on the electric field strength  $E$  indicate that, at a certain value of the critical field (at a temperature  $T = 300$  K,  $E_{cr} = 800$ , and 41 and 40 V/cm for TlS, TlSe and TlInS<sub>2</sub>, respectively), the Tl<sup>+</sup> ion sub-lattice may undergo step wise disordering, which is accompanied by a step wise change in conductivity. We find it possible that the applied field causes disordering (melting) of the cationic sub-lattice, which leads to a

stepwise increase in the occupancy of interstitial sites throughout the crystal volume.

### 3. Conclusion

Our studies demonstrate that the temperature dependences of the conductivity of TlS, TlSe, and TlInS<sub>2</sub> crystals indicate the dominant character of ionic conductivity above the critical temperatures. We explain this effect through the sharp change in the number of ions in the states characterized by high ion mobility, i.e., by the phase transition to the superionic state. It is remarkable that for all three crystals this phase transition occurs approximately at 380 K. We have evaluated the values of the activation energy for ionic conductivity in TlS, TlSe, and TlInS<sub>2</sub> crystals.

The electric field dependences of the conductivity reveal the effect of S type switching in these crystals. This effect can be explained by the stepwise increase in the occupancy of interstitial sites throughout the crystal volume under the applied electric field.

In our opinion, due this effect, TlS, TlSe, and TlInS<sub>2</sub> crystals could be considered promising materials for fabricating nano-size switches.

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